ENANTIOSELECTIVE TOTAL SYNTHESIS AND PHARMACOLOGIC PROFILE OF 12-DEOXY-12(S)-METHYL LEUKOTRIENE B_4

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Abstract

An enantioselective synthesis of a unique analogue of the pro-inflammatory eicosanoid Leukotriene B₄ has been achieved in a convergent manner, from readily available starting materials. This analogue, 12-deoxy-12(S)-methyl LTB₄, provides convincing evidence that the 12(R)-hydroxy group present in the natural ligand is essential for its chemoattractant properties towards inflammatory cells, such as neutrophils.

Leukotriene B₄ [(5S,12R) dihydroxy-6,14-cis, 8,10-trans-eicosatetraenoic acid] (LTB₄),¹ has been shown to stimulate the aggregation and degranulation of human neutrophils, promote chemotaxis and chemokinesis of leukocytes and is a mediator of lysosomal enzyme release and superoxide generation.² It has been postulated to be a major inflammatory mediator in a number of disease states such as psoriasis and inflammatory bowel disease (IBD).³

As part of our studies directed towards the identification of LTB₄ antagonists based on the structure of the natural ligand itself ⁴ we chose 12-deoxy-12(S)-methyl LTB₄ as an attractive target in light of our belief that the 12-hydroxy group of the natural ligand was critical for receptor mediated signal transduction via a guanine nucleotide-binding protein.⁵

Leukotriene B₄ R=OH 12-Deoxy-12(S)-methyl LTB₄ R=CH₃

12-Deoxy-12(S)-methyl Leukotriene B₄

Reagents

a) nickel boride, H₂, 90% b) (COCl)₂, then n-BuLi/(4R,5S)-(+)-4-Methyl-5-phenyl-2-oxazolidinone, THF, -78% to -10°C, 99%, C)NaHMDS, CH₃I, -78°C to 0°C, 80%, d) LiBH₄/H₂O (1:1), Et₂O, 85%, e) (COCl)₂, DMSO, Et₃N, f) $\underline{6}$, LiHMDS, THF, -25°C, 88%, g) H₂, Lindlar, 99%, h) \underline{n} -Bu₄NF, THF, RT

The synthesis of the desired target was accomplished as in Scheme I. Starting with the readily available 4-decynoic acid 1,6 we utilized Evans' innovation for the asymmetric alkylation of chiral imide enolates7 to introduce the key 12(S) stereochemistry required in the natural product analog. Carboxylic acid, 1 was reduced to the corresponding (Z)-olefin in 90% yield and then converted uneventfully to the chiral acyloxazolidinone using Evans' conditions. Metallation and quenching of the enolate derived from 2 with methyl iodide afforded the α-methyl acyloxazolidinone 3 with diastereoisomeric excess of ≥ 97% and chemical yield of 80%. Reductive cleavage of the chiral auxiliary with lithium borohydride/water, in the manner recently described by us,8 followed by Swern oxidation produced enantiomerically pure aldehyde 4. This material was coupled with the Wadsworth-Emmons reagent 6 according to the procedure of Nicolaou et all in 88% yield after chromatographic purification. Lindlar reduction (99% yield) followed by desilylation with tetra nbutylammonium fluoride in THF and concomitant ester hydrolysis furnished 12-deoxy-12(S)methyl LTB_{4.10} This synthesis once again demonstrates the considerable utility of Evans' acyloxazolidinone chemistry in the synthesis of LTB4 or 12-hydroxy eicosatetraenoic acid analogs,11 and that the lithium borohydride/H₂0 reagent system is a very effective agent for the nondestructive reductive removal of sterically encumbered acyloxazolidinone auxiliaries.

12-Deoxy-12(S)-methyl-LTB₄ was evaluated in a series of pharmacologic assays designed to establish its ability to function as an antagonist of the natural ligand itself. ¹² Although the compound demonstrated an IC₅₀ of 3.8 μ M in a LTB₄ human neutrophil receptor binding assay, it was devoid of either agonist or antagonist properties in a modified Boyden chamber chemotaxis assay at doses ranging from 10-5 to 10-8M. Interestingly enough, however, it displayed a dose related inhibition of human neutrophil degranulation induced by LTB₄ with an IC₅₀ of 3.4 μ M.

This observation lends strong support to the current hypothesis that there are two functionally or even structurally different, high and low affinity LTB₄ receptors on human neutrophils; a high affinity site associated with the chemotactic response and a low affinity site associated with enzyme secretion and degranulation and suggests that these sites can be differentially regulated by pharmacologic agents such as 12-deoxy-12(S)-methyl LTB₄.13

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- Satisfactory analytical data eg PMR, CMR, IR, microanalysis and/or high resolution mass 10. spectrum were obtained for all samples. Key data includes, 12-deoxy-12(S)-Methyl LTB4 1HNMR (CDCl₃,5OO MHz) 86.39 (1H,dd, J=15,10Hz) 8-H, 6.22 (1H,dd,J=15,10Hz) 9-H, 6.08 (1H,dd,J=15, 10Hz) 10-H, 6.06 (1H,t, J=11Hz) 7-H, 5.71 (1H,dd,J=15,7.5Hz) 11-H, 5.41 (1H, m) (dt, J=10, 7Hz), 15-H, 5.38 (1H, m) 6-H, 5.32 (1H, m) (dt, J=10, 7 Hz) 14-H, 4.60 (1H, q, J=8 Hz) 5-H, 2.38 (2H, m) 2Hs, 2.26 (1H, heptet, J=7Hz) 12-H, 2.06 (2H, t, J=7Hz) 13-Hs, 2.00 (2H, q, J=7Hz) 16-Hs, 1.4-1.8 (4H, complex band) 3 and 4 Hs, 1.2-1.4 (6H, complex band) 17, 18, 19-Hs, 1.01 (3H, d, J=7Hz) 12-CH₃ and 0.88 (3H, t, J=7Hz) 20-CH₃ [8-H corresponds to proton attached to C-8 of LTB₄ analog - LTB₄ numbering system], IR (CHCl₃) 3250 (br.OH), 3000,2937,2833,2812,1700 (C=0),1550,1450,1400,1330,1225, 1150,1050,1015,970,760 cm⁻¹. Alcohol derived from LiBH₄ reduction of 3 ¹H NMR (CDCl₃, 300 MHz), δ5.4 (2H,m), 3.5 (2H,m), 2.0 (4H, m), 1.7 (2H, m), 1.3 (7H, m), 0.9 (6H,m). IR (CHCl₃) 3010,2960,2920,2860,1450,1360,1280,1255,1230,1210,1200,1130, 1100,1000 cm-1, $[\alpha]_{n}^{25}$ =+3.4 (c 1.25, CHCl₃), 98% ee as determined by HPLC analysis of the Mosher esters.
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